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terms in this set 28 the chemical formula for an ionic compound represents the of the combined ions present in a sample ratio a formula that shows only the types and numbers of atoms combined in a single molecule is called a n molecular formula

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if one knows the mass and molar mass of reactant a and the molar mass of product d in a chemical reaction one can determine the mass of product d produced by using the a mole ratio of d to a from the chemical equation b group numbers of the elements of a and d in the periodic table

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