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igsce chemistry answers b the mass of the empirical formula is 4 12 8 1 16 72 since this is equal to the relative formula mass the molecular formula is the same as the empirical formula that is  $\text{c}_4\text{h}_8\text{o}_{10}$  you know the mass of anhydrous sodium sulfate 1 42 g you can work out the mass of water of crystallisation

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